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~~identifying types of chemical bonds from~~  
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~~Ionic vs. Molecular~~

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Polar Covalent Bonds and Nonpolar  
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exam. **Chemical Bonds: Covalent vs.**

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AND COMPOUNDS | IONIC ,

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COMPOUNDS Covalent Bonding |

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A chemical bond is formed when electrons are shared between two atoms. There are three types of bonds: covalent bonds, polar covalent bonds and ionic bonds. The simplest example of bonding is demonstrated by the  $H_2$  molecule. We can see that each hydrogen atom has a single electron from the periodic table.

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12-17 Main content: Chemical Bonds

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7.2 IONIC BONDING The chemical bond

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formed by transfer of electron from a metal to a non- metal is known as ionic or electrovalent bond . For example, when sodium metal and chlorine gas are brought into contact, they react violently and we obtain sodium chloride. This reaction is shown below:  $2\text{Na(s)} + \text{Cl}_2\text{(g)} \rightarrow 2\text{NaCl(s)}$



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## *7 CHEMICAL BONDING*

There are many types of chemical bonds that can form, however the 3 main types are: ionic, covalent, and metallic bonds. You must become familiar with how they work and the differences between the 3 types. I. Ionic bonding: Model 1 is a description of what chemists call ionic

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bonding. Ionic bonding occurs strictly between metal and nonmetal atoms.

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Accordingly, there are different types of chemical bonds, like, Ionic or electrovalent bond Covalent bond Coordinate covalent bond In addition to these we have a special kind of bond called hydrogen bond. Let us discuss about

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different types of bonds, their formation and the properties of the compounds so formed. 5.2 Ionic or Electrovalent Bond

## *CHEMICAL BONDING*

There are two main types and some secondary types of chemical bonds: 1  
Ionic bond Ionic bonding involves a

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transfer of an electron, so one atom gains an electron while one atom loses an electron. One of the resulting ions carries a negative charge (anion), and the other ion carries a positive charge (cation).

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